Total Synthesis of the Rubrolone Aglycon

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Abstract: A total synthesis of the rubrolone aglycon is detailed and is based on two key Diels–Alder reactions. The AB ring system incorporating a tetrasubstituted pyridine was assembled, enlisting the rare 4π participation of an *O*-alkyl α,β -unsaturated oxime in an intramolecular [4 + 2] cycloaddition reaction (70%). The C-ring oxygenated tropolone was introduced through a room-temperature, exo selective [4 + 2] cycloaddition of a cyclopropenone ketal (97%) followed by in situ generation of a norcaradiene and room-temperature electrocyclic rearrangement to a cycloheptatrienone ketal appropriately substituted for hydrolysis directly to a 2,4-dihydroxycycloheptatrienone.

Rubrolone (1),¹ a red tropoloalkaloid isolated from *Strepto-myces enchinoruber*, was identified in a single-crystal X-ray structure determination and shown to possess the unique azuleno[2,3-*c*]pyridine-2,5,13-trione aglycon **2** characteristic of a class of structurally related agents (Figure 1).^{2,3} To date, only a single total synthesis of the rubrolone aglycon has been reported⁴ despite this unique structure. As part of the exploration of the Diels—Alder reactions of azadienes^{5–11} and the thermal cycloaddition reactions of cyclopropenone ketals,^{12–19} their respective potential for introduction of the AB and C rings of rubrolone have been examined.^{3b,20} Herein we describe the

(1) Schuep, W.; Blount, J. F.; Williams, T. H.; Stempel, A. J. Antibiot. 1978, 31, 1226.

(2) Palleroni, N. J.; Reichelt, K. E.; Mueller, D.; Epps, R.; Tabenkin, B.; Bull, D. N.; Schuep, W.; Berger, J. *J. Antibiot.* **1978**, *31*, 1218.

(3) AB ring system: (a) Kelly, T. R.; Liu, H. J. Am. Chem. Soc. **1985**, 107, 4998. (b) Boger, D. L.; Zhu, Y. Tetrahedron Lett. **1991**, 32, 7643.

(4) Aglycon total synthesis: Kelly, T. R.; Echavarren, A.; Whiting, A.; Weibel, F. R.; Miki, Y. *Tetrahedron Lett.* **1986**, *27*, 6049.

(5) Boger, D. L.; Weinreb, S. M. Hetero Diels-Alder Methodology in Organic Synthesis; Academic Press: San Diego, 1987. Boger, D. L. Tetrahedron **1983**, 39, 2869. Boger, D. L. Chemtracts: Org. Chem. **1996**, 9, 149.

(6) Boger, D. L.; Corbett, W. L.; Curran, T. T.; Kasper, A. M. J. Am. Chem. Soc. **1991**, 113, 1713 and references therein. Boger, D. L.; Curran, T. T. J. Org. Chem. **1990**, 55, 5439. Boger, D. L.; Corbett, W. L.; Wiggins, J. M. J. Org. Chem. **1990**, 55, 2999.

(7) Hwang, Y. C.; Fowler, F. W. J. Org. Chem. 1985, 50, 2719.

(8) Ito, Y.; Nakajo, E.; Saegusa, T. Synth. Commun. 1986, 16, 1073.

(9) Serckx-Poncin, B.; Herbain-Frisque, A.-M.; Ghosez, L. *Tetrahedron Lett.* **1982**, *23*, 3261. For reports of the failure of α , β -unsaturated oximes and their derivatives to act as 4π components in [4 + 2] cycloaddition reactions, see ref 5, 6, and references therein.

(10) Ihara, M.; Kirihara, T.; Kawaguchi, A.; Fukumoto, K.; Kametani, T. *Tetrahedron Lett.* **1984**, *24*, 4541.

(11) Teng, M.; Fowler, F. W. J. Org. Chem. 1990, 55, 5646.

(12) Boger, D. L.; Brotherton-Pleiss, C. E. In *Advances in Cycloaddition Chemistry*; Curran, D. P., Ed.; JAI Press: Greenwich, CT.; Vol. 2, pp 147– 219.

(13) Boger, D. L.; Brotherton, C. E. J. Am. Chem. Soc. **1986**, *108*, 6695. (14) Boger, D. L.; Brotherton, C. E. Tetrahedron **1986**, *42*, 2777.

(15) Colchicine: Boger, D. L.; Brotherton, C. E. J. Am. Chem. Soc. 1986, 108, 6713. Boger, D. L.; Brotherton, C. E. J. Org. Chem. 1985, 50, 3425.
 Grandirubrine and imerubrine: Boger, D. L.; Takahashi, K. J. Am. Chem. Soc. 1995, 117, 12452.

(16) Boger, D. L.; Wysocki, R. J., Jr. J. Org. Chem. 1989, 54, 714.

(17) Boger, D. L.; Brotherton, C. E. J. Am. Chem. Soc. 1984, 106, 805.

(18) Boger, D. L.; Brotherton, C. E. *Tetrahedron Lett.* **1984**, 25, 5611.

(19) Boger, D. L.; Brotherton, C. E.; Georg, G. I. Tetrahedron Lett. 1984, 25, 5615.





realization of a total synthesis, based on these two unusual Diels-Alder reactions, of the rubrolone aglycon 2. Key to the approach was the implementation of a rare 4π participation of an O-alkyl α,β -unsaturated oxime in an intramolecular Diels-Alder reaction for construction of the tetrasubstituted pyridine with assemblage of the AB ring system (Scheme 1, $11 \rightarrow 12$) and the Diels-Alder reaction of the cyclopropenone ketal 17 with the highly oxygenated diene 16 as a prelude to C-ring tropolone introduction. Conversion of the [4 + 2] cycloadduct 18 to the norcaradiene 29, in situ low-temperature electrocyclic rearrangement to a cycloheptatrienone ketal and tautomerization was anticipated to provide a fully oxygenated precursor to the rubrolone aglycon 2. Inherent in the design of the tropolone annulation was the incorporation of three oxygen substituents in the diene-dienophile reaction partners, permitting the direct preparation of a 2,4-dihydroxycycloheptatrienone in a process complementary to those we have detailed in total syntheses of grandirubrine/imerubrine and colchicine based on the [4 + 2]and [3 + 4] cycloaddition reactions of cyclopropenone ketals, respectively.12-15

Synthesis of the Rubrolone AB Ring System: 4π Participation of an *O*-Alkyl α_{β} -Unsaturated Oxime in an Intramolecular Diels—Alder Reaction. The AB ring system 12 was prepared by an approach we previously disclosed^{3b} (Scheme 2). Condensation of aldehyde 3^{21} with 1-lithio-1-pentyne provided 4 (90%). Protection of the secondary alcohol as the

⁽²⁰⁾ Model rubrolone ABC ring system: Boger, D. L.; Zhu, Y. J. Org. Chem. 1994, 59, 3453.

⁽²¹⁾ Nicolaou, K. C.; Prasad, C. V. C.; Hwang, C.-K.; Duggran, M. E.; Veale, C. A. J. Am. Chem. Soc. **1989**, 111, 5321.



THP ether **5** (DHP, PPTS, CH₂Cl₂, 99%) followed by removal of the OTBS ether (Bu₄NF, THF, 99%) and PDC oxidation of the primary alcohol **6** (77%) provided aldehyde **7**. Treatment of **7** with dimethyl (2-oxopropyl)phosphonate provided the key intermediate **8** in 96% yield. Condensation of **8** with *O*-methyl hydroxylamine hydrochloride (pyridine, 96%) followed by removal of the THP-protecting group of **9a** (Amberlyst, MeOH, 96%) provided **10a**. Subsequent Swern oxidation of **10a** afforded **11a** (86%), which provided the initial *O*-alkyl α , β -unsaturated oxime for study. Similarly, treatment of **8** with *O*-benzyl hydroxylamine hydrochloride provided **9b** (96%). Removal of the THP-protecting group of **9b** (99%) followed by Swern oxidation of **10b** (95%) provided **11b**, an additional oxime for study.

Table 1 summarizes representative results obtained from a study of the intramolecular [4 + 2] cycloaddition of **11**. No reaction was found to occur at temperatures lower than 140 °C, and a productive rate of reaction was observed at 170–200 °C (triisopropylbenzene, bp 233–236 °C). The elimination of methanol (**11a**) or benzyl alcohol (**11b**) with formal oxidation





of the initial cycloadduct to the corresponding pyridine was found to occur under the reaction conditions to provide 12 directly in 70% yield. Precedent for this Diels-Alder reaction may be found in the use of α,β -unsaturated N,N-dimethylhydrazones in intermolecular⁹ and intramolecular²² [4 + 2]cycloaddition reactions, even though the general reported failure of α,β -unsaturated oxime cycloadditions suggested that their use may not prove viable.9 Since our initial disclosure,3b additional examples of such reactions have been disclosed.^{23,24} Thus, the observation that the O-alkyl α . β -unsaturated oximes 11 participate as effective 4π components of an intramolecular Diels-Alder reaction with an electron-deficient dienophile indicate that the introduction of an alkoxy electron-donating substituent (OR) on the nitrogen atom of the inherently electrondeficient 1-aza-1,3-butadiene system, like the dimethylamino group of the unsaturated dimethylhydrazones (NMe₂), may be sufficient to promote its participation in a normal, HOMO_{diene}controlled Diels-Alder reaction.

Synthesis of the Rubrolone Aglycon: [4 + 2] Cycloaddition of a Cyclopropenone Ketal. In a prior study, we demonstrated that the 1,4-addition of 2-bromoinden-1-one with

⁽²²⁾ Dolle, R. E.; Armstrong, W. P.; Shaw, A. N.; Novelli, R. Tetrahedron Lett. 1988, 29, 6349.

⁽²³⁾ Kusurkar, R. S.; Bhosale, D. K. *Tetrahedron Lett.* **1991**, *32*, 3199. Snyder, S.; Vosburg, D. A.; Jarvis, M. G.; Markgraf, J. H. *Tetrahedron* **2000**, *56*, 5329.

⁽²⁴⁾ Vijn, J. R.; Arts, J. H.; Green, R.; Castelijns, A. M. *Synthesis* **1994**, *13*, 573. Behforouz, M.; Gu, Z.; Stelzer, L. S.; Ahmadian, M.; Haddad, J.; Scherschel, J. A. *Tetrahedron Lett.* **1997**, *38*, 2211.

Scheme 3



the higher-order cyanocuprate²⁵ prepared from 2-lithio-1,4dioxene was followed by the elimination of HBr to provide the corresponding diene which could serve as a suitably functionalized substrate for a Diels–Alder approach to rubrolone C-ring introduction.²⁰ However, all attempts to extend this approach to the preparation of **16** from **12** failed to afford the desired diene. Nonetheless, the diene **16** was obtained by Stille coupling of **14** with **15** which is formally equivalent to such a conjugate addition. Thus, C6 hydroxylation of **12** with PhI(OAc)₂²⁷ (KOH, MeOH), under conditions that provide the corresponding α -hydroxy dimethyl ketal in a reaction that first generates the aviodoso intermediate and is followed by the formation of the alkoxy epoxide and ring opening of the epoxide with a MeOH trap, followed by elimination of water ((CF₃CO)₂O, Et₃N, 65%) gave **13**, Scheme 3.

Bromination of **13** (Br₂, CH₂Cl₂) and subsequent elimination of HBr (*t*-BuOK, DMF, 91% for two steps) selectively afforded **14** which is derived from an E2 elimination with removal of the most acidic and sterically most accessible hydrogen. Stille



Figure 2.

coupling of **14** with **15**²⁶ cleanly provided the diene **16** in superb yield upon treatment with $(Ph_3P)_4Pd$ (DMF-THF 1:4, 95%). No coupling product was observed when Pd(II) catalysts such as bis(benzonitrile)palladium dichloride were used, and a mixture of DMF-THF (1:4) as solvent gave **16** in high yield while the use of DMF or THF individually resulted in decreased yields.

The key Diels–Alder reaction of **16** with the cyclopropenone ketal **17**²⁸ was conducted at room temperature and was complete within 45 min, providing the single cycloadduct **18** in 97% yield. The success of this rapid and unusually effective [4 + 2] cycloaddition reaction may be attributed to the combined use of a reactive electron-rich diene and the strained dienophile **17**. The [4 + 2] cycloaddition reaction provided a single diastereomer, which could be purified by simple crystallization from EtOAc/hexanes. Single-crystal X-ray structure analysis²⁹ revealed that **18** was the anticipated cycloadduct shown in Supporting Information Figure 1a derived from exclusive cycloaddition through the less sterically encumbered exo transition state^{14,15} (Figure 2).

Treatment of 18 with NBS-MeOH provided 19 in 80% yield as a single diastereomer, the structure of which was established by single-crystal X-ray structure analysis.²⁹ The stereochemical outcome indicates that the initial reaction of 18 with NBS proceeded selectively on the less hindered face (Supporting Information Figure 1b). Selective deprotection of dimethyl ketal was accomplished by brief treatment with aqueous TFA (CH2-Cl₂, quant.) to give ketone **20**. Elimination of HBr from **20** was effected by mild treatment with DBU, and was followed by acid-catalyzed hydrolysis of the mixed ketal (90% aqueous TFA, CH₂Cl₂, 72% for two steps) to provide 21. In light of the number of potentially competitive reactions, this transformation proceeded smoothly and cleanly to 21. Base treatment of the corresponding bromohydrin, derived from the reaction of the [4 + 2] cycloadduct 18 with NBS-H₂O, failed to give the analogous elimination product and resulted instead in the formation of the remarkable and stable epoxide 25 (Scheme 4)

⁽²⁵⁾ Blanchot-Courtois, V.; Hanna, I. *Tetrahedron Lett.* 1992, *33*, 8087.
(26) Blanchot-Courtois, V.; Fetizon, M.; Hanna, I. *Synthesis* 1990, *9*, 755.

⁽²⁷⁾ Moriarty, R. M.; Hu, H.; Gupta, S. C. *Tetrahedron Lett.* **1981**, *22*, 1283.

⁽²⁸⁾ Isaka, M.; Ejiri, S.; Nakamura, E. Tetrahedron 1992, 48, 2045.

⁽²⁹⁾ The author has deposited the atomic coordinates for these structures with the Cambridge Crystallographic Data Centre, and these are allocated the deposition numbers (18: CCDC 147871, 25: CCDC 147872, 19: CCDC 147873). The coordinates may be obtained upon request from the Director, Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge, CB2 1 EZ, UK.

Scheme 4



whose structure was established by X-ray analysis²⁹ (Supporting Information Figure 1c). Although not extensively investigated, treatment of either **19** or **25** with base under various conditions (*t*-BuOK, KOH, or DBU) gave no evidence of desired elimination product. Thus, ketone activation for elimination of HBr from **20** was required, and the subsequent formation of the cycloheptatrienone derivative **21** is derived from a reaction sequence that proceeds by mixed ketal hydrolysis ($26 \rightarrow 27$), enolization with norcaradiene generation ($28 \rightarrow 29$), electrocyclic rearrangement to the cycloheptatrienone ketal ($29 \rightarrow 30$), and tautomerization to **21** (Scheme 5).

Subjection of **21** to hydrolysis (LiOH, THF–MeOH–H₂O 3:1:1, 99%) led to clean monodeprotection with selective generation of **22**. The remaining 2-hydroxyethyl ether of **22** could not be removed by either vigorous acid or base-catalyzed hydrolysis and was removed in a two-step sequence involving conversion of the alcohol to the primary bromide **23** (TMSBr, MeCN, 99%)³⁰ followed by reductive cleavage of the 2-bromoethyl ether (Zn, NH₄Cl, EtOH). Under these conditions, the benzylic ketone is also reduced, leading to the generation of **24**. Without purification or optimization, this crude alcohol was oxidized (NBS, DMSO) to provide **2** in 48% yield (two steps from **23**) completing the preparation of the rubrolone aglycon.

Experimental Section

2-Methyl-5-oxo-4-propyl-6,7-dihydro-5*H***-cyclopenta**[*c*]**pyridine** (12). A solution of 11a³¹ (298 mg, 1.58 mmol) in triisopropylbenzene

(35 mL) was stirred under Ar at 185 °C for 36 h. The reaction mixture was purified directly by column chromatography (SiO₂, 2×30 cm, 30% EtOAc-hexane) to give **12**³¹ (132 mg, 70%) as white needles: mp 56–57 °C (EtOAc-hexane).

5,5-Dimethoxy-2-methyl-4-propyl-5H-cyclopenta[*c*]**pyridine (13).** A solution of **12** (1.30 g, 6.87 mmol) in MeOH (70 mL) was treated with KOH (3.85 g, 68.7 mmol) and PhI(OAc)₂ (4.43 g, 13.7 mmol), and the reaction mixture was stirred at 0 °C for 3 h. The reaction mixture was diluted with Et₂O (500 mL), and the mixture was washed with saturated aqueous NaCl (300 mL), dried (Na₂SO₄), filtered, and concentrated in vacuo. The residue was taken up in CH₂Cl₂ (70 mL) and treated with trifluoroacetic anhydride (1.94 mL, 13.7 mmol) and Et₃N (4.77 mL, 34.4 mmol), and the reaction mixture was stirred at 25 °C for 24 h. The reaction mixture was diluted with CH₂Cl₂ (200 mL), and the organic phase was washed with saturated aqueous NaCl (100 mL), dried (Na₂SO₄), filtered, and concentrated in vacuo. Flash chromatography (SiO₂, 5 × 15 cm, 1% Et₃N and 20% EtOAc—hexane) afforded **13**³¹ (1.04 g, 65%) as a light tan syrup.

7-Bromo-5,5-dimethoxy-2-methyl-4-propyl-5*H*-cyclopenta[*c*]pyridine (14). A solution of 13 (60 mg, 0.25 mmol) in CH₂Cl₂ (2.5 mL) was treated with Br₂ (79 mg, 0.5 mmol), and the reaction mixture was stirred at 0 °C for 2 h. The volatiles were removed in vacuo. The residue in DMF (2.5 mL) was treated with *t*-BuOK (1.0 M solution in THF, 0.75 mL, 0.75 mL), and the reaction mixture was stirred at 0 °C for 2 h. The reaction mixture was diluted with EtOAc (10 mL), and the organic phase was washed with H₂O (10 mL) and saturated aqueous NaCl (10 mL), dried (Na₂SO₄), filtered, and concentrated in vacuo. Flash chromatography (SiO₂, 1 × 6 cm, 1% Et₃N and 20% EtOAc—hexane) afforded 14³¹ (71 mg, 91%) as a light tan syrup.

7-(1,4-Dioxen-2-yl)-5,5-dimethoxy-2-methyl-4-propyl-5H-cyclopenta[*c*]**pyridine (16).** A solution of **14** (53 mg, 0.17 mmol) and **15**²⁶ (76 mg, 0.20 mmol) in DMF–THF (1:4, 1.7 mL) was treated with (Ph₃P)₄Pd (209 mg, 0.19 mmol), and the reaction mixture was stirred at 50 °C for 2 h. The reaction mixture was diluted with EtOAc (20 mL), and the mixture was washed with H₂O (20 mL) and saturated aqueous NaCl (10 mL), dried (Na₂SO₄), filtered, and concentrated in vacuo. Flash chromatography (SiO₂, 1 × 8 cm, 1% Et₃N and 20% EtOAc–hexane) afforded **16**³¹ (51 mg, 95%) as a light tan syrup.

Cycloadduct 18. Compound **16** (350 mg, 1.10 mmol) was treated with **17**²⁸ (403 mg, 2.88 mmol) at 25 °C for 45 min. Crystallization of the mixture from EtOAc-hexane afforded **18**³¹ (420 mg, 84%) as colorless prisms. The filtrate was concentrated in vacuo, and flash chromatography of the mother liquors (SiO₂, 1 × 5 cm, 1% Et₃N and 33% EtOAc-hexane) afforded additional **18** (66 mg, 13%; 97% total) as a white solid: mp 164–166 °C (EtOAc-hexane).

Compound 19. A solution of **18** (280 mg, 0.61 mmol) in MeOH– THF (1:1, 5 mL) was treated with NBS (163 mg, 0.92 mmol), and the reaction mixture was stirred at 0 °C for 10 min before aqueous $Na_2S_2O_3$ and saturated aqueous $NaHCO_3$ (50 mL) were added. The reaction mixture was extracted with EtOAc (3 × 50 mL), and the organic layers were washed with saturated aqueous NaCl (30 mL), dried (Na_2SO_4), filtered, and concentrated in vacuo. Crystallization of the residue from EtOAc-hexane afforded **19**³¹ (280 mg, 80%) as colorless prisms: mp 153–155 °C (EtOAc-hexane).

Compound 20. A solution of **19** (100 mg, 0.17 mmol) in 20 mL of CH₂Cl₂ was treated with 90% aqueous TFA (500 μ L), and the reaction mixture was stirred at 25 °C for 10 min before saturated aqueous NaHCO₃ (30 mL) was added. The mixture was extracted with CH₂Cl₂ (50 mL), and the organic layer was washed with saturated aqueous NaCl (30 mL), dried (Na₂SO₄), filtered, and concentrated in vacuo to afford **20**³¹ (88 mg, quantative) as a colorless oil which was used without further purification.

2-[(2,2-Dimethyl-3-hydroxypropyl)oxy]-5,13-dioxo-9-methyl-11propyl-4-[(2-hydroxyethyl)oxy]azuleno[2,3-c]pyridine (21). A solution of 20 (88 mg, 0.17 mmol) in 2 mL of CH₂Cl₂ was treated with DBU (31 μ L, 2.0 mmol), and the reaction mixture was stirred at 0 °C for 10 min. The mixture was diluted with CH₂Cl₂ (20 mL), and the

⁽³⁰⁾ Treatment of **24** with NBS and PPh₃ gave only a trace of **25**, whereas NBS/PBu₃ effectively provided **25** although it was difficult to isolate free of reaction byproducts. Bates, H. A.; Farina, J.; Tong, M. *J. Org. Chem.* **1986**, *51*, 2637.

⁽³¹⁾ Full details of the preparation and characterization of 4-11 and full characterization data for 12-14, 16, 18-23, and 2 are provided in the Supporting Information.

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organic layer was washed with saturated aqueous NaCl (10 mL), dried (Na₂SO₄), filtered, and concentrated in vacuo. A solution of the residue in 2 mL of CH₂Cl₂ was treated with 90% aqueous TFA (200 μ L), and the reaction mixture was stirred at 25 °C for 1 h before addition of saturated aqueous NaHCO₃ (30 mL). The mixture was extracted with CHCl₃ (30 mL), and the organic layer was washed with saturated aqueous NaCl (10 mL), dried (Na₂SO₄), filtered, and concentrated in vacuo. Flash chromatography (SiO₂, 1 × 4 cm, 2% MeOH–CHCl₃) afforded **21**³¹ (52 mg, 72%) as a yellow powder.

5,13-Dioxo-2-hydroxy-9-methyl-11-propyl-4-[(2-hydroxyethyl)oxy]azuleno[2,3-c]pyridine (22). A suspension of **21** (21.3 mg, 0.05 mmol) in 2 mL of THF-MeOH-H₂O(3:1:1) was treated with LiOH-H₂O (10.5 mg, 0.25 mmol), and the reaction mixture was stirred at 50 °C for 2 h. The mixture was cooled to 0 °C, and the THF was removed under a stream of N₂. The yellow precipitate was collected by filtration to afford **22**³¹ (17.0 mg, 99%) as a yellow powder.

5,13-Dioxo-2-hydroxy-9-methyl-11-propyl-4-[(2-bromoethyl)oxy]azuleno[2,3-c]pyridine (23). A suspension of **22** (4.0 mg, 0.012 mmol) in 2 mL of CH₃CN was treated with TMSBr (23 μ L, 0.18 mmol), and the reaction mixture was warmed at reflux for 3 h. The mixture was cooled and washed with hexane (3 × 4 mL), and the CH₃CN layer was concentrated in vacuo to afforded **23**³¹ (4.9 mg, quantitative) as a yellow solid sufficiently pure to use directly in the next reaction.

Rubrolone Aglycon (2). A solution of **23** (2.0 mg, 4.9 μ mol) in 1 mL of EtOH was treated with Zn (3.2 mg, 49.6 μ mol) and NH₄Cl (2.7

mg, 49.6 μ mol), and the reaction mixture was stirred at 60 °C for 3 h. The reaction mixture was filtered, and the filtrate was concentrated in vacuo. A solution of the residue containing **24** in 0.5 mL of DMSO was treated with NBS (5.5 mg, 24.5 μ mol), and the reaction mixture was stirred at 25 °C for 1 h before H₂O was added. The reaction mixture was concentrated in vacuo. Flash chromatography (reverse phase, 1 × 4 cm, 20% H₂O–MeOH) afforded **2**³¹ (0.7 mg, 48%) as an orange solid identical in all respects with properties reported for authentic material.⁴

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Supporting Information Available: Full experimental details and characterization for 4-11 and full characterization data for 12-14, 16, 18-23 and 2, and Supporting Information Figure 1a-c (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

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